INTRODUCTION

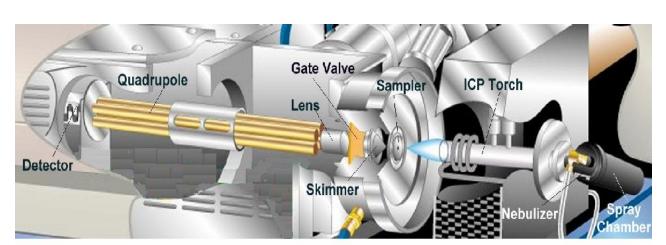
The determination of major, minor and trace elements in samples has always been of paramount importance in the geochemical exploration of elements of economic interest and the evaluation of potential resources. However, these types of samples pose an unusual set of challenges for the analyst, including: the range of analyte concentrations present; the number of elements required per sample; and the volume of samples produced by a typical exploration survey. These factors are further compounded by the presence of a complex range of matrices and the potential for widely different compositions within an analytical batch of samples.

Over the past decade, inductively coupled plasma mass spectrometry (ICP-MS) has pushed the boundaries of both detection limits and the range of determinable elements available to the exploration geologist, whilst addressing the criteria outlined above. This has developed hand in hand with an increasingly sophisticated understanding of elemental behaviour and the development of software for handling multi-element arrays, leading to the use of a large suite of elements as pathfinders for mineralization.

Initial reconnaissance samples (or soil samples) generally require a partial extraction, followed by an elemental scan, to provide information concerning anomalies above background levels. The aim of this work was to utilize the multi-element capability of the ICP-MS for samples from a case study carried out at the Namex Explorations Inc. Huffman Lake Property (Ontario, Canada) using the MMI-M technique, in order to highlight the ability of ICP-MS to provide unparalleled analytical data for the geologist.

Inductively Coupled Plasma Mass Spectrometry (ICP-MS)

ICP-MS utilizes an argon plasma as an efficient heat source for the formation of positive ions. A small portion of the ions enters the instrument, is focused by the lens and passes into the quadrupole mass spectrometer, where ions are separated based on their mass to charge ratio. An electron multiplier is used to detect the ions as they pass out of the mass spectrometer. By comparing intensities from known standard solutions to the unknown sample, the concentration of the analyte can be determined.



The Inductively Coupled Plasma Mass Spectrometer

The Dual Stage Detector allows for the determination of trace levels of elements in solution using the pulse stage, and higher concentrations using the analogue stage. By cross-calibrating the two stages an increased dynamic range can be achieved. For example, Pb concentrations measured using a maximum calibration of 0.050mg/L have been shown to be linear from 0.0001mg/L up to 10mg/L.

The main advantages of ICP-MS for geological analysis are: •The sensitivity of the technique allows for the determination of many elements in the parts per billion range •The multi-element capability allows for the analysis of a wide range of elements at different concentration levels in one pass

Mobile Metal Ion (MMI) Analysis by ICP-MS

Partial weak extractions, such as Mobile Metal Ion Geochemistry (MMI) have become a significant tool in geochemical exploration over the past decade (Mann et. al., 1998). Earlier versions of these leaches measured four or five indicator elements at low concentrations using ICP-MS in order to determine locations of buried gold and base metal mineralization. The recent introduction of MMI-M, a pH neutral extraction that aids in the identification of polymetallic targets and multi-element associations for specific styles of mineralization, has placed the emphasis on ICP-MS to deliver quantitative results for a much larger range of elements. As well as measuring base and precious metals, MMI-M has been used in the determination of rare earth elements (significant in the lithogeochemistry of kimberlite and nickel sulphide deposits) and major elements such as Fe, Ca and Mg. In total, ICP-MS can be used to measure at least 45 elements (see Table 1) present in concentrations ranging from sub parts per billion (e.g. Au and Th), through to hundreds of parts per million (e.g. Mg and Ca).

Element	D.L.	Element	D.L.	Element	D.L.
Ag	0.001	Fe	1	Sm	0.001
AI	1	Gd	0.001	Sn	0.001
As	0.01	La	0.001	Sr	0.01
Au	0.0001	Li	0.005	Та	0.001
Ba	0.01	Mg	1	Tb	0.001
Bi	0.001	Мо	0.005	Те	0.01
Ca	10	Nb	0.0005	Th	0.0005
Cd	0.001	Nd	0.001	Ti	0.003
Ce	0.005	Ni	0.005	TI	0.0005
Со	0.005	Pb	0.01	U	0.001
Cr	0.1	Pd	0.001	W	0.001
Cu	0.01	Pr	0.001	Y	0.005
Dy	0.001	Rb	0.005	Yb	0.001
Er	0.0005	Sb	0.001	Zn	0.02
Eu	0.0005	Sc	0.005	Zr	0.005

Table 1. Element list and Detection Limits (D.L.) for the Multi element MMI-M analysis. All detection limits in mg/kg.

Reference: Mann, A.W., Birrell, R.D., Mann, A.T., Humphreys, D.B. and Perdix, J.L., 1998. Application of mobile metal ion technique to routine geochemical exploration. J. Geochem. Explor. 61:87-102

METHODS & MATERIALS

50g of sample was shaken for 20min with 50mL of MMI-M Selective Weak Extraction Solution (Wamtech, Australia). The solution was allowed to stand overnight and analysed by ICP-MS (Elan 9000, PerkinElmer/SCIEX, Concord, Canada). The instrument was calibrated using a blank, 0.010mg/L (0.001mg/L for Ag, Au, Pd and 1mg/L for Al, Ca, Fe, Mg) and 0.050mg/L (0.005mg/L for Ag, Au, Pd and 5mg/L for Al, Ca, Fe, Mg) standards, all containing the MMI-M matrix.

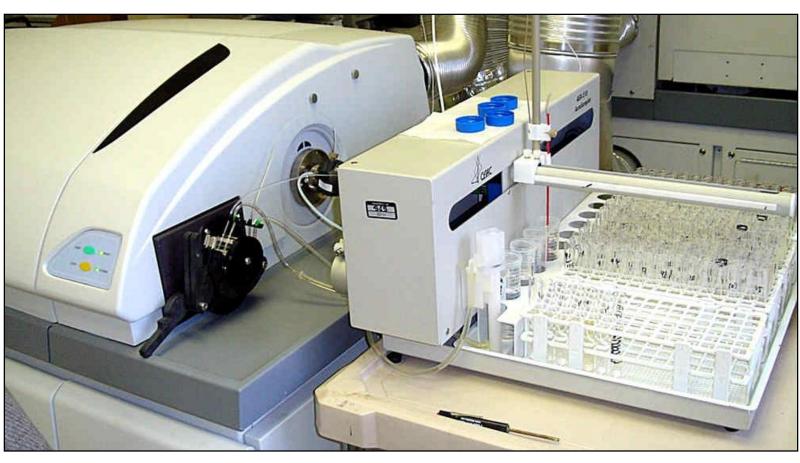
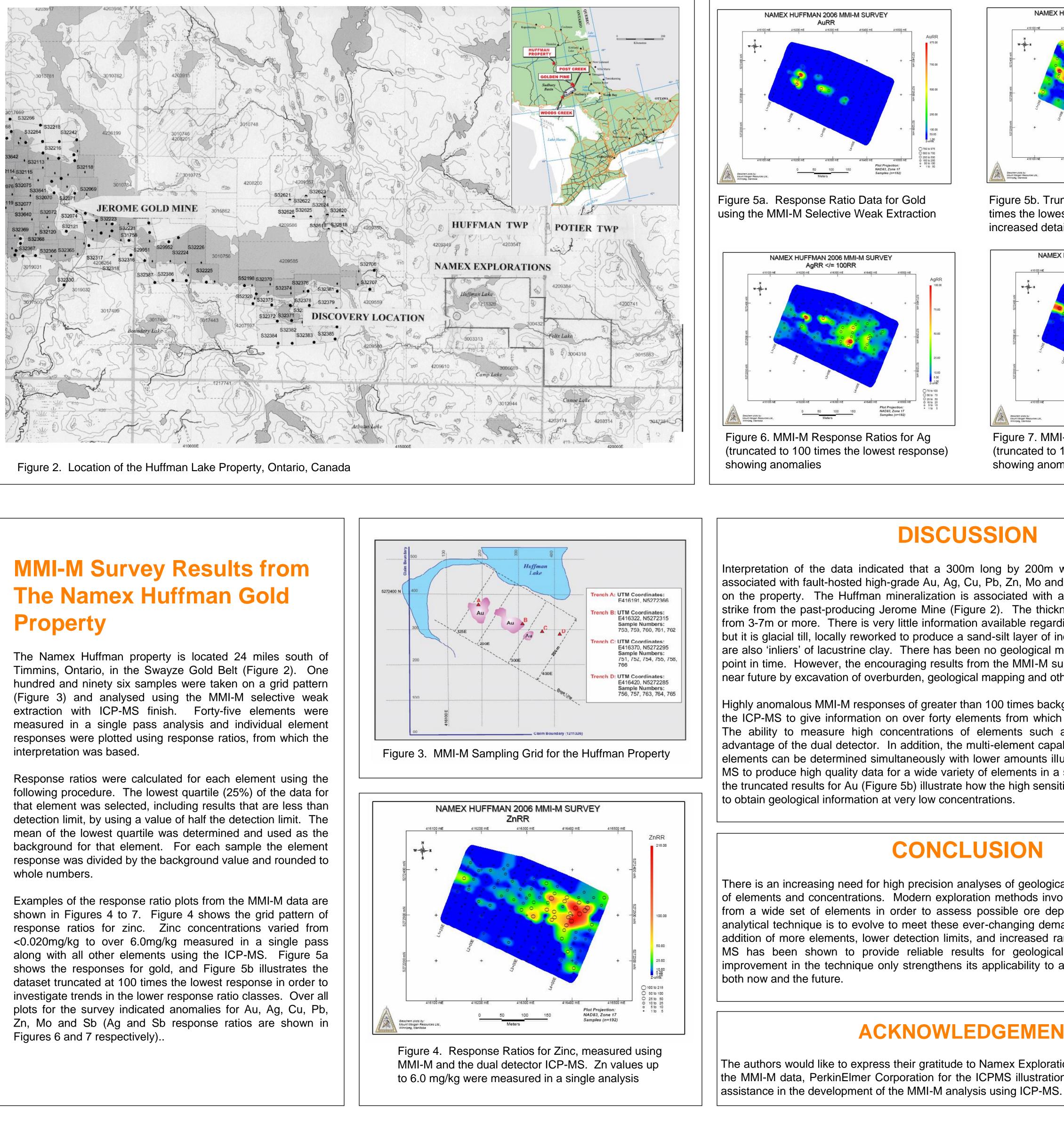


Figure 1. Analysis of MMI-M Extracts using ICP-MS

The Evaluation of Geological Exploration Samples using Multielement Mobile Metal Ion (MMI-M) Selective Weak Extraction and Inductively Coupled Plasma Mass Spectrometry (ICP-MS)

Nicholas Turner^{[1]*}, Dana Mills^[1], Mark Fedikow^[2] & Pierrette Prince^[1] ^[1] SGS Minerals Services, 1885 Leslie Street, Toronto, ON, M3B 2M3, Canada. ^[2] Namex Explorations Inc., Canada.



*Nicholas.Turner@sgs.com

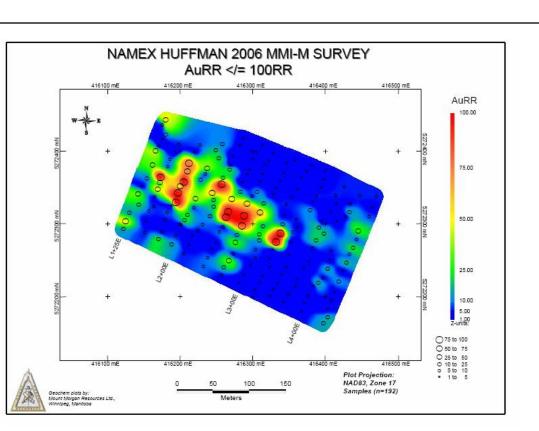


Figure 5b. Truncated Results for Gold at 100 times the lowest response ratio, showing increased detail of Au anomalies

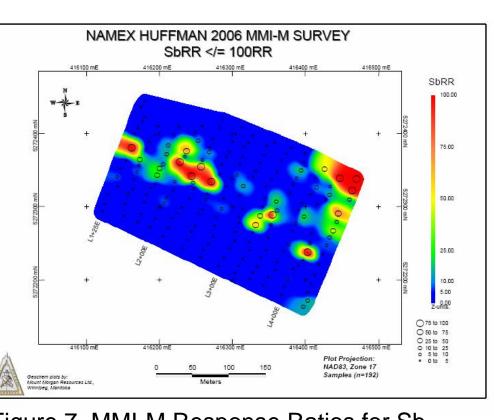


Figure 7. MMI-M Response Ratios for Sb (truncated to 100 times the lowest response) showing anomalies

DISCUSSION

Interpretation of the data indicated that a 300m long by 200m wide geochemical anomaly is associated with fault-hosted high-grade Au, Ag, Cu, Pb, Zn, Mo and Sb mineralization that occurs on the property. The Huffman mineralization is associated with a felsic porphyry and is along strike from the past-producing Jerome Mine (Figure 2). The thickness of the overburden varies from 3-7m or more. There is very little information available regarding the age of the overburden but it is glacial till, locally reworked to produce a sand-silt layer of indeterminate thickness. There are also 'inliers' of lacustrine clay. There has been no geological mapping on the property to this point in time. However, the encouraging results from the MMI-M survey will be followed up in the near future by excavation of overburden, geological mapping and other investigative techniques.

Highly anomalous MMI-M responses of greater than 100 times background were quantitated using the ICP-MS to give information on over forty elements from which the conclusions were drawn. The ability to measure high concentrations of elements such as Zn (Figure 4) shows the advantage of the dual detector. In addition, the multi-element capability, where the high levels of elements can be determined simultaneously with lower amounts illustrates the power of the ICP-MS to produce high quality data for a wide variety of elements in a single analysis. Furthermore, the truncated results for Au (Figure 5b) illustrate how the high sensitivity of ICP-MS can be utilized

CONCLUSION

There is an increasing need for high precision analyses of geological samples for a greater range of elements and concentrations. Modern exploration methods involve statistical analysis of data from a wide set of elements in order to assess possible ore deposits. The challenge for the analytical technique is to evolve to meet these ever-changing demands. These can include: the addition of more elements, lower detection limits, and increased ranges of concentrations. ICP-MS has been shown to provide reliable results for geological samples and the constant improvement in the technique only strengthens its applicability to a wide variety of situations for

ACKNOWLEDGEMENT

The authors would like to express their gratitude to Namex Explorations Inc. for permission to use the MMI-M data, PerkinElmer Corporation for the ICPMS illustration and to Alan Mann for all his